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The nitro compounds are one class of important industrial reagents and typically are prepared by reducing nitrite compounds, typically potassium nitrite, with hydrogen in the presence of catalysts. As a class, the nitro compounds are not water-soluble, and hence present handling and disposal problems. In the gas phase industry, nitric oxide is produced as a by-product of various chemical processes such as the chemical reduction of nitrogen oxide containing by-products. Nitric oxide is a valuable intermediate for the manufacture of nitrogen based fertilizers and pharmaceuticals, polymer precursors, and specialty chemicals such as, for example, alkali metal nitrites. The use of nitric oxide in making nitrogen based fertilizers and pharmaceuticals is described in an article titled "Nitric oxide from molecular nitrogen" by J. R. Barnett et al. which appears in *Analytical Chemistry*, Vol. 65, Number 17 (1993). Examples of how nitric oxide is reacted to form nitrogen-containing polymers are provided in U.S. Pat. Nos. 5,213,906; 5,569,748; 5,296,162; 5,374,512; 5,399,600; 5,405,974; 5,437,990; 5,554,374; 5,668,251; 5,750,629; 5,981,751; 5,994,494; 6,093,744 and 6,444,739. Nitric oxide has found use in pharmaceuticals, pesticides and specialty chemical intermediates. In U.S. Pat. No. 5,119,147, nitric oxide is prepared from a nitric oxide containing gas and a metal nitrite, which metal nitrite is normally present as the nitrite

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salt of a strong mineral acid, such as ammonium nitrite, and produced in a liquid phase from a solution of the nitrite. A nitric oxide is produced at a temperature in the range of from  $-15^{\circ}\text{C}$ . to  $40^{\circ}\text{C}$ . with a molecular oxygen source in the presence of a catalyst. Another nitric oxide producing process is described in U.S. Pat. No. 3,299,058 which describes the production of nitric oxide from a nitric oxide containing gas and oxygen by a gas phase reaction of the nitric oxide with oxygen in the presence of a catalyst in which the product gas effluent containing

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